Acta Crystallographica Section E

Structure Reports Online

ISSN 1600-5368

2-Bromo-1-(2,4-dichloro-5-fluorophenyl)-3-phenylprop-2-en-1-one

V. Dhanasekaran, D. Gayathri, D. Velmurugan, K. Ravikumar and M. S. Karthikeyan .

^aDepartment of Crystallography and Biophysics, University of Madras, Guindy Campus, Chennai 600 025, India, ^bLaboratory of X-ray Crystallography, Indian Institute of Chemical Technology, Hyderabad 500 007, India, and ^cDepartment of Chemistry, Mangalore University, Mangalore 574 199, India

Correspondence e-mail: d_velu@yahoo.com

Key indicators

Single-crystal X-ray study $T=293~\mathrm{K}$ Mean $\sigma(\mathrm{C-C})=0.004~\mathrm{\mathring{A}}$ R factor = 0.030 wR factor = 0.077 Data-to-parameter ratio = 17.8

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

The molecular geometry of the title compound, $C_{15}H_8BrCl_2FO$, is stabilized by a $C-H\cdots Br$ interaction.

Received 27 March 2007 Accepted 2 April 2007

Comment

Chalcone derivatives have nonlinear optical (NLO) properties with good blue light transmittance. They also have a wide range of biological applications, such as anticancer (Rao *et al.*, 2004) and anti-inflammatory activity (Hsieh *et al.*, 2000). In view of the optical and medicinal importance of chalcone derivatives, we have prepared the title compound.

The geometric parameters of (I) lie within their expected ranges (Allen *et al.*, 1987). The dihedral angle between the two benzene rings is $89.1 (2)^{\circ}$. The O1-C7-C8-Br1 torsion angle is $17.4 (4)^{\circ}$. The molecular conformation is stabilized by a $C-H\cdots Br$ interaction (Table 1).

Experimental

1-(2,4-Dichloro-5-fluorophenyl)-3-phenylprop-2-en-1-one (1 mmol) was prepared by a literature procedure (Shivarama Holla *et al.*, 2006). To a solution of 1-(2,4-dichloro-5-fluorophenyl)-3-phenylprop-2-en-1-one) (1 mmol) in chloroform (25 ml), bromine (1 mmol) was added slowly with stirring. After the completion of addition of bromine (1 mmol), the reaction mixture was stirred for 24 h. Excess of chloroform was distilled off and precipitated 2,3-dibromo-1-(2,4-dichloro-5-fluorophenyl)-3-phenylpropan-1-one was filtered off and dried. A mixture of dibromopropanone (1 mmol) and triethylamine (1 mmol) in dry benzene (30 ml) was stirred for 24 h. The excess of solvent when removed under reduced pressure gave the title compound. It was crystallized from acetone by slow evaporation.

Crystal data

 $C_{15}H_8BrCl_2FO$ $M_r = 374.02$ Monoclinic, Pc a = 6.5423 (8) Å b = 14.3317 (16) Å c = 7.8040 (9) Å $\beta = 93.284$ (2)° V = 730.52 (15) Å³ Z = 2Mo $K\alpha$ radiation $\mu = 3.18$ mm⁻¹ T = 293 (2) K $0.23 \times 0.22 \times 0.21$ mm

© 2007 International Union of Crystallography All rights reserved

organic papers

Data collection

Bruker SMART APEX CCD areadetector diffractometer Absorption correction: none 7433 measured reflections 3234 independent reflections 2946 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.036$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.030$ $wR(F^2) = 0.077$ S = 0.933234 reflections 182 parameters 2 restraints H-atom parameters constrained $\Delta \rho_{\rm max} = 0.27 \ {\rm e \ \mathring{A}^{-3}}$ $\Delta \rho_{\rm min} = -0.45 \ {\rm e \ \mathring{A}^{-3}}$ Absolute structure: Flack (1983), 1526 Friedel pairs Flack parameter: 0.049 (9)

Table 1
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D-\mathrm{H}\cdots A$
C11-H11···Br1	0.93	2.74	3.319 (3)	122

The H atoms were positioned geometrically and were treated as riding on their parent C atoms, with C—H = 0.93 Å and with $U_{\rm iso}({\rm H})$ = $1.5 U_{\rm eq}({\rm C})$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 2003); software used to prepare material for publication: *SHELXL97* and *PARST* (Nardelli, 1995).

VD thanks the Department of Science and Technology (DST) for providing a fellowship. DV thanks DST for a major project.

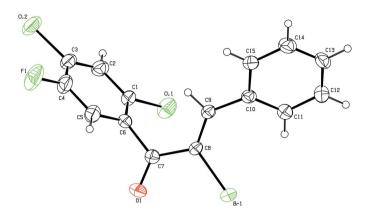


Figure 1

The molecular structure of the title compound showing 30% probability displacement ellipsoids.

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans.* 2, pp. S1–19.

Bruker (2001). SAINT (Version 6.28a) and SMART (Version 5.625). Bruker AXS Inc., Madison, Wisconsin, USA.

Flack, H. D. (1983). Acta Cryst. A39, 876-881.

Hsieh, H.-K., Tsao, L.-T., Wang, J.-P. & Lin, C.-N. (2000). *J. Pharm. Pharmacol.* **52**, 163–171.

Nardelli, M. (1995). J. Appl. Cryst. 28, 659.

Rao, Y. K., Fang, S.-H. & Tzeng, Y.-M. (2004). Bioorg. Med. Chem. 12, 2679–2686.

Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Shivarama Holla, B., Sooryanarayana Rao, B., Sarojini, B. K., Akberali, P. M. & Suchetha Kumari, N. (2006). *Eur. J. Med. Chem.* **41**, 657–663.

Spek, A. L. (2003). J. Appl. Cryst. 36, 7–13.